ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.003 Å R factor = 0.043 wR factor = 0.138 Data-to-parameter ratio = 12.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

5-Amino-1-(1,5-dimethyl-1*H*-pyrazol-4-ylcarbonyl)-3-methylsulfanyl-1*H*-1,2,4-triazole

In the title compound, $C_9H_{12}N_6OS$, the pyrazole and triazole rings are nearly coplanar, forming a dihedral angle of $6.50~(9)^\circ$. There are $N-H\cdots N$ intermolecular hydrogen-bond interactions in the crystal structure, providing stabilization. Received 1 March 2005 Accepted 30 March 2005 Online 9 April 2005

Comment

Many pyrazole and triazole derivatives have been reported to show various biological activities, such as antifungal (Chen & Li, 2000), herbicidal (Ren *et al.*, 2000), insecticidal (Huang *et al.*, 1996) and other activities (Kopp *et al.*, 2001). Thus, we paid special attention to the possibility of obtaining a pyrazole ring connected to a triazole ring *via* a carbonyl group. In order to develop new biological activities, we synthesized the title compound, (I), the structure of which is reported here.



Bond distances and angles (Table 1) are as expected for this type of compound. The pyrazole and triazole rings are nearly coplanar, the dihedral angle between them being 6.50 (9)°. A weak intramolecular $N-H\cdots O$ hydrogen-bond interaction is observed (Table 2). In the crystal structure, centrosymmetrically related molecules are linked in dimers through the formation of intermolecular $N-H\cdots N$ hydrogen-bond interactions (Table 2).

Experimental

A mixture of 1,5-dimethylpyrazol-4-ylcarbonyl hydrazide (3 mmol) and CIDT (*N*-cyanoimido-*S*,*S*-dimethylthiocarbonate) (2 mmol) in acetonitrile (15 ml) was refluxed for 8 h (monitored by thin-layer chromatography) until a solid product formed; the solution was cooled and the product filtered off. The pure product was isolated by recrystallization from dimethylformamide (m.p. 514 K).

Crystal data	
C ₉ H ₁₂ N ₆ OS	$D_x = 1.458 \text{ Mg m}^{-3}$
$M_r = 252.31$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 2811
a = 7.642(5) Å	reflections
b = 10.100 (7) Å	$\theta = 2.4-27.8^{\circ}$
c = 15.250 (10) Å	$\mu = 0.28 \text{ mm}^{-1}$
$\beta = 101.275 \ (8)^{\circ}$	T = 293 (2) K
$V = 1154.3 (13) \text{ Å}^3$	Prism, colourless
Z = 4	$0.59 \times 0.38 \times 0.20 \text{ mm}$

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organic papers



Figure 1

View of the title compound, with 35% probability ellipsoids.



Bruker APEX II CCD area-	2024 independent reflections
detector diffractometer	1763 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\rm int} = 0.020$
Absorption correction: multi-scan	$\theta_{\rm max} = 25.0^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -9 \rightarrow 8$
$T_{\min} = 0.882, \ T_{\max} = 0.946$	$k = -12 \rightarrow 10$
5982 measured reflections	$l = -18 \rightarrow 18$
Refinement	
Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.088P)^2]$

+ 0.3815P] where $P = (F_o^2 + 2F_c^2)/3$

 $\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$

 $\Delta \rho_{\rm min}$ = -0.50 e Å⁻³

 $(\Delta/\sigma)_{\rm max} = 0.001$

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.043$ $wR(F^2) = 0.138$ S = 1.082024 reflections 158 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

O1-C6	1.221 (3)	N5-C9	1.313 (3)
N1-C4	1.336 (3)	N5-C7	1.376 (3)
N2-C1	1.320 (3)	N6-C9	1.333 (3)
N3-C9	1.388 (3)	C1-C5	1.412 (3)
N4-C7	1.307 (3)	C4-C5	1.395 (3)
C4-N1-N2	113.28 (17)	N2-C1-C5	111.94 (19)
C1-N2-N1	104.29 (17)	N1-C4-C5	106.22 (18)
C9-N3-N4	108.46 (15)	C4-C5-C1	104.27 (19)
C7-N4-N3	101.66 (16)	N4-C7-N5	116.75 (19)
C9-N5-C7	103.03 (18)	N5-C9-N3	110.08 (17)

Table 2

H	ſyd	lrogen-	bonding	geometry ((A	۱ , °).
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$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\begin{array}{c} N6-H6B\cdots O1\\ N6-H6A\cdots N5^{i} \end{array}$	0.89	2.23	2.695 (3)	112
	0.89	2.09	2.961 (3)	163

Symmetry code: (i) -x, 1 - y, 1 - z.



Figure 2

The molecular packing of the title compound, viewed along the a axis. Hydrogen bonds are shown as dashed lines.

All H atoms were placed in calculated positions, with C-H = 0.93 or 0.96 Å and N-H = 0.89 Å, and included in the final cycles of refinement using a riding model, with $U_{\rm iso}({\rm H})$ set at $1.2U_{\rm eq}({\rm C})$ for CH₂, and $1.5U_{\rm eq}({\rm C},{\rm N})$ for NH, CH and CH₃ H atoms.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

This project was supported by the Natural Science Foundation of Shandong Province (No. Y2003B01).

References

Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA. Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison,

Wisconsin, USA.

Chen, H. S. & Li, Z. M. (2000). Chin. J. Chem. 18, 596-602.

 Huang, R. Q., Song, J. & Feng, L. (1996). Chem. J. Chin. Univ. 17, 1089–1091.
Kopp, M., Lancelot, J. C., Dallemagne, P. & Rault, S. (2001). J. Heterocycl. Chem. 38, 1045–1050.

Ren, T. R., Yang, H. W., Gao, X., Yang, X. L. & Zhou, J. J. (2000). Pest. Manag. Sci. 56, 218–226.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.